D-LACTATE DEHYDROGENASE

from Microorganism

PREPARATION and SPECIFICATION

Appearance: White amorphous powder, lyophilized

Activity : 220U/mg-solid or more

Contaminants : GOT and GPT all $\leq 5.0 \times 10^{-3}\%$

Malate dehydrogenase $\leq 1.0 \times 10^{-2} \%$ NADH oxidase $\leq 1.0 \times 10^{-3} \%$ Myokinase $\leq 1.0 \times 10^{-2} \%$ Pyruvate Kinase $\leq 1.0 \times 10^{-3} \%$

PROPERTIES

Stability : Product shipped on dry ice, but long-term storage should be at -20°C.

Molecular weight : 39 kDa **Isoelectric point** : 6.1

Michaelis constant : 6.41×10-4M (Pyruvate, pH 7.0)

Inhibitors : Co^{2+} , Fe^{3+} , Cu^{2+} , Hg^{2+} , Ag^+ , SDS

 Optimum pH
 : 7.0
 (Fig.1)

 Optimum temperature
 : 40°C
 (Fig.2)

 pH stability
 : pH 4.5~10.5 (25°C, 20hr)
 (Fig.3)

Thermal stability: below 50°C (pH 7.0,15 min) (Fig.4)

Effect of various : (Table 1)

chemicals

UNIT DEFINITION

One unit causes the oxidation of one micromole of NADH per minute at pH 7.0 at 25°C

APPLICATIONS

This enzyme is useful for enzymatic determination of numerous metabolites, e.g., ATP, ADP, glucose, creatinine, pyruvate, lactate and glycerol, and of enzyme activities, e.g., GPT, PK, and CPK when coupled with the related enzymes.

Manufactured in an ISO 9001 certified facility: Suzhou SignalChem Biotechnologies Corp.

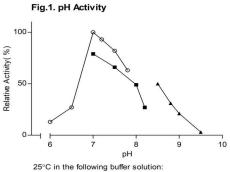


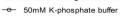
Table 1. Effect of Various Chemicals on D-Lactate dehydrogenase

The enzyme was dissolved in 0.05M K-phosphate buffer, pH 7.5 containing 0.1% of BSA (40U/ml) and incubated with each chemical at 25°C for 1hr.

Chemical	Concn.(mM)	Residual activity(%
None	_	100
CaCl ₂	2	94
MgSO ₄	2	95
ZnSO ₄	2	86
NiCl ₂	2	89
CoCl ₂	2	58
$MnCl_2$	2	96
FeCl ₃	2	63
CuSO ₄	2	78
AgNO ₃	2	6
HgSO ₄	2	1
NEM	2	80
IAA	2	89

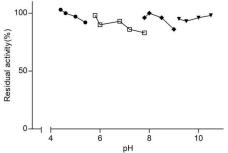
Chemical	Concn.(mM)	Residual activity(%)
BME	2	91
Hydroxylamine	2	93
EDTA	5	95
NaF	20	90
NaN ₃	20	96
Borate	50	89
Proclin-300	0.045% (v/v)	90
SDS	0.05%	47
Na-Cholate	0.1%	97
Tween-20	0.1% (v/v)	96
Triton X-100	0.1% (v/v)	97
Span-20	0.1% (v/v)	94
Brij-35	0.1%	94





^{-■ 50}mM Tris-HCl buffer

^{▲ 50}mM Glycine-NaOH buffer



25°C, 20hr-treatment with following buffer solution:

0.1M Acetate buffer

O.1M K-phosphate buffer

→ 0.1M Tris-HCl buffer

→ 0.1M Glycine-NaOH buffer



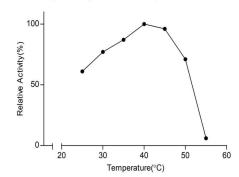
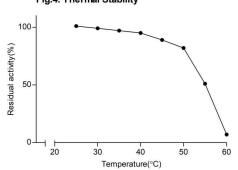


Fig.4. Thermal Stability



15min- treatment with 50mM K-phosphate buffer, pH7.0